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الملف Worksheet about Acid and base salt

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CONTENT STANDARD 6.1 THE ROLE OF WATER IN SHOWING ACIDIC AND ALKALINE PROPERTIES

Acid is _____ that _____ in water to produce _____.

Example : $\text{HCl} \rightarrow$ _____ + _____ $\text{HNO}_3 \rightarrow$ _____ + _____

Hydrogen ion will combine with water molecules to form _____ ion.

Basicity of acid

_____ produce one hydrogen ion per acid molecule.

_____ produce two hydrogen ion per acid molecule.

_____ produce three hydrogen ion per acid molecule.

Base and Alkali

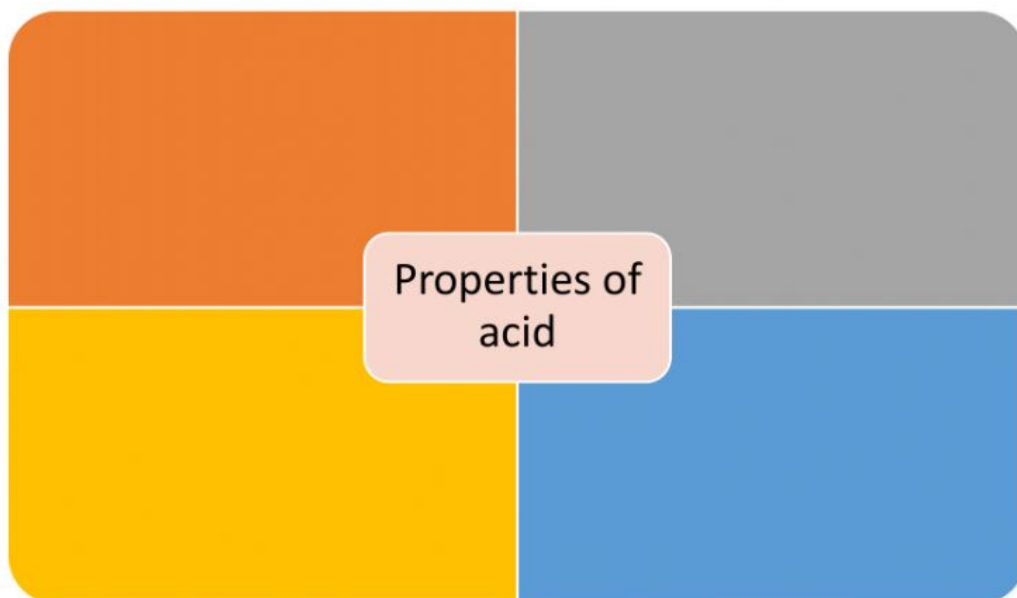
Base is substance that react with _____ to produce _____ and _____ only

Example $\text{MgO} + \text{H}_2\text{SO}_4 \rightarrow$ _____ + _____

Alkali is _____ that _____ in water to produce _____.

Example : $\text{NH}_3 + \text{H}_2\text{O} \rightarrow$ _____ + _____

Properties of acid and alkali





Properties of
alkali