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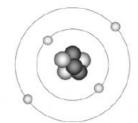
## Table Periodic about Worksheet الملف

<u>Almanahj Website</u>  $\rightarrow$  <u>American curriculum</u>  $\rightarrow$  <u>5th Grade</u>  $\rightarrow$  <u>Chemistry</u>  $\rightarrow$  <u>Term 1</u>  $\rightarrow$  <u>The file</u>

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Name:	Period:
1. Which of the following sets nucleus of an atom?	contains objects where BOTH are found in the

- A. electrons and protons
- B. neutrons and electrons
- C. protons and energy
- D. protons and neutrons
- 2. An atom consists of 11 electrons, 12 neutrons, and an undetermined number of protons. Which information will be MOST helpful in determining the number of protons and the atom's overall charge?
  - A. mass number
  - B. conductivity
  - C. number of energy levels
  - D. chemical family
- 3. What element is represented in this model if the atomic number is 3?
  - A. lithium
  - B. carbon
  - C. nitrogen
  - D. neon



- 4. Which of the following lists the elements in order, from those having the fewest protons to those having the greatest number of protons?
  - A. Li, Be, He, C, N
  - B. N, P, As, Sb, Bi
  - C. Cs, Rb, Po, At, Rn
  - D. At, I, Br, Cl, F

1	H	2	
2	3 LI	4 Be	
3	11 Na	12 <b>Mg</b>	3
4	19 K	20 Ca	21 Se
5	37 Rb	38 Sr	39 Y
5	55 Cs	56 Ba	57 La
- 1			720

7 Fr Ra Ac

	13	14	15	16	17	He
	5 B	6 C	7 N	8	9 F	10 Ne
12	43	14	15	16	17	18
	Al	Si	P	S	CI	Ar
30	31	32	33	34	35	35
Zn	Ga	Ge	As	Se	Br	Kr
48	49	50	51	52	53	54
Cd	In	Sn	Sb	Te	1	Xe
80	B1	82	83	84	85	86
Hg	TI	Pb	Bi	Pe	At	Rn

Name:	Period:
	ed into groups and periods of elements. The up of elements are listed below.
<ul> <li>Low boiling and melting</li> </ul>	noints
7 valence electrons	s points
Which of the following element	ts is a member of this group?
A. Boron	
B. Bromine	
C. Magnesium	

6. An atom of a certain element has 27 protons, 27 electrons, and a mass number of 59. At room temperature this element is a shiny, silver-gray metal. How many neutrons are in this atom?

7. Which statement is true about electrons?

D. Sulfur

- A. Electrons contribute most of the mass of an atom.
- B. Electron placement determines an atom's reactivity.
- C. Electron placement determines properties such as luster.
- D. Electrons determine whether an atom is liquid at room temperature.