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Covalent and bonds Ionic and forces Intermolecular and forces Intramolecular about Worksheet الملف

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Learning Target: I can describe and explain the differences between intramolecular and intermolecular forces.



## Intermolecular vs. Intramolecular Forces Video Notes

Ι.	Why are electrons attracted to th	e nucleus of an atom?		
2.	Why are cations and anions attra-	cted to each other?		
nt	ermolecular vs. Intramolecular Fo	rces:		
	Intramolecular forces are attracti	ve forces that hold		
	(or substance). These are	and usually called	bonds.	
	Intermolecular forces are	attractive forces that exist between		

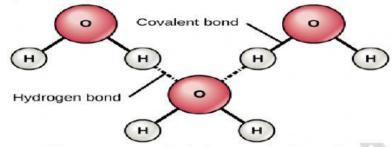
3. Label the parts of the diagram below where the intramolecular and intermolecular forces are.

$$\delta^{\dagger} \quad \delta^{-}$$
 $H - CI$ 
 $\delta^{\dagger} \quad \delta^{-}$ 
 $H - CI$ 
 $H - CI$ 

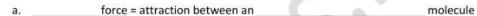
Created By: Chivas & Jordan Spivey

Learning Target: I can describe and explain the differences between intramolecular and intermolecular forces.

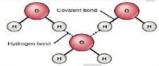
4. Label the parts of the diagram below where the intramolecular and intermolecular forces are.



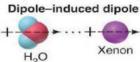
- 5. What are some of the strongest intramolecular forces in order of strength?
  - a. bonds = attraction between
  - b. bonds = attraction between
  - c. Bonds = attraction between
- 6. What are some of the strongest intermolecular forces in order of strength?



b. \_\_\_\_\_\_ bond = attraction between special type of \_\_\_\_\_



- c. Forces = attraction between  $H CI = \delta^+ \delta^- \delta^- \delta^-$
- d. \_\_\_\_\_ Dipole = attraction between \_\_\_\_\_ molecules.



e. \_\_\_\_\_Forces = attraction between \_\_\_\_\_

Review:

- Intramolecular forces are attractive forces that exist between atoms within

- Intermolecular forces are attractive forces that exist between

- The common types of intramolecular forces, known as chemical bonds are

- The common types of intermolecular forces, ranked roughly by strength, are \_\_\_\_\_