نموذج اختبار تجريبي وفق الهيكل الوزاري منهج انسباير متبوع بالإجابات





تم تحميل هذا الملف من موقع المناهج الإماراتية

موقع المناهج ← المناهج الإماراتية ← الصف الخامس ← علوم ← الفصل الأول ← ملفات متنوعة ← الملف

تاريخ إضافة الملف على موقع المناهج: 11-19-2025 08:45

ملفات ا كتب للمعلم ا كتب للطالب ا اختبارات الكترونية ا اختبارات ا حلول ا عروض بوربوينت ا أوراق عمل منهج انجليزي ا ملخصات وتقارير ا مذكرات وبنوك ا الامتحان النهائي ا للمدرس

المزيد من مادة علوم:

إعداد: School Aasimah Al

التواصل الاجتماعي بحسب الصف الخامس











صفحة المناهج الإماراتية على فيسببوك

الرياضيات

اللغة الانجليزية

اللغة العربية

التربية الاسلامية

المواد على تلغرام

المزيد من الملفات بحسب الصف الخامس والمادة علوم في الفصل الأول	
حل تدريبات نهائية وفق الهيكل الوزاري منهج انسباير	1
تدريبات نهائية وفق الهيكل الوزاري منهج انسباير بدون الحل	2
أسئلة الامتحان النهائي القسم الورقي منهج انسباير المسار المتقدم	3
حل كراسة تدريبية مراجعة وفق الهيكل الوزاري الجديد منهج بريدج	4
مذكرة مراجعة أسئلة وفق الهيكل الوزاري الجديد منهج بريدج	5



نهاية اختبار نهاية الفصل الأول التجريبي 2025 / 2026 End of Term 1 Mock Exam 2025/2026

Student Number			
Student Name	ahi		
School	1030	To M	
Class	0/		
Stream	026	General	
Cubicat	6:	Science	
Subject		Inspire	
	قم الك	liel	

PARTS	Total	Marks
1. Multiple Choice	60	
2. Writing	40	
TOTAL	100	

ركِّز على امتحان العلوم وابذل أقصى ما لديك.

PART 1—MULTIPLE CHOICES

Question				1
How does matter change from one state to another?				
Α	remov	ing energy	С	Both A and B are correct
B adding energy		D	adding mass	

Q	uestion	ahi		2
Liquid ha	as	- ana	1.60	2
А	A Definite volume		С	No definite volume
В	B Definite shape		D	All are correct

				2025	
Question			3		
Which o	Which of the following statements is correct?				
А		eds to be solid to be natter.	С	Matter can be a solid, liquid or gas.	
В	Matter can be	e a solid or a liquid.	D	Matter can be a solid, liquid or gas but does not include living things.	

Qu	estion	4		
Why is an inflated balloon having more volume than a bag of marbles?				
Α	Because	it has more space	С	Because it has mass.
В	Because it co	ntains more particles.	D	All are correct.

Qı	uestion			5
Which comparison between gases and solids is correct?				
А	•	cannot move while ids can.	С	Gas particles are farther apart than solid particles.
В	Gas particles are closer together than solid particles.		D	Gas particles have a fixed shape while solids do not.

Question				6	
Which be	Which best explains why gases can be compressed more easily than solids and liquids?				
А	Gas particles are very far apart with lots of empty space.		С	Gas particles cannot move at all.	
В	Gas particles are tightly packed together.		D	Gas particles are heavier than solids.	

Qu	estion		7				
Which of	Which of the following is NOT a chemical change?						
А	Combustion	a la	Boiling				
В	Digestion	D	Rusting				

Question				8	
You combine 500 g of one material and 200 g of another in a closed container. A chemical chang occurred. After the chemical change, the total mass recorded was 700 g. What law explains this					
А	A Law of Conservation of Mass			Law of Gravity	
В	Law of Energy		D	Law of Nature	

Question 9

A spoonful of salt has a mass of 10 grams. A cup of water has a mass of 300 grams. What is the total mass of the saltwater when the salt is dissolved in the water?



А	300 grams		More than 300 grams
В	Less than 300 grams	D	All are correct

Question				10
Which mixture is most likely a solution?			J. CO	
А	Cranberry juice		С	Potting soil
В	Muddy water		D	Milk

Question				11
Which of the following physical properties can help identify copper?				ntify copper?
А	(Color	С	Conductivity
В	Ref	lectivity	D	All are correct

Question		12			
Other than the change in color, which of the following can tell that a chemical change happened?					
А	Temperature change		С	Formation of solid	
В	Formation of gas		D	All are correct	

Question				13
Why does sugar dissolve faster in warm water?				
А	Warm water is sweet.		С	Warm water changes sugar's color.
В	Cold water makes sugar bigger.		D	Warm water moves faster and breaks sugar faster.

Question	14

Look at the diagram of the Sun, the Earth and the Moon What would a night time observer on Earth see?







А	No moon	С	A crescent moon
В	A quarter moon	D	A full moon

Planet	Length of Day (hours)	Length of Year (Earth years)	Distance from the Sun (AU)
Mercury	1,408	0.2	0.4
Venus	5,832	0.6	0.7
Earth	24	1.0	1.0
Mars	25	1.9	1.5
Jupiter	10	11.9	5.2
Saturn	10	29.4	9.5
Uranus	17	84.0	19.2
Neptune	16	164.8	30.0

A The farther a planet is from the Sun, the longer its day. B Uranus is the coldest planet. C A day on Earth is longer than a day on Venus. The farther a planet is from the Sun, the longer its year.

Question		16		
How does the moon's gravity cause tides on			arth?	
А	It pushes the water away from Earth.		С	It makes the sun hotter.
В	It changes the color of the ocean.		D	It pulls on Earth's water, making high and low tides.

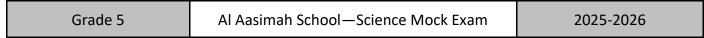
Q	Question			17
Why is fo	olding a paper a p	ohysical change?		
А	The paper changes shape but not its material.		С	No new substance is formed.
В	The paper can be unfolded back to its original form.		D	All answers are correct.

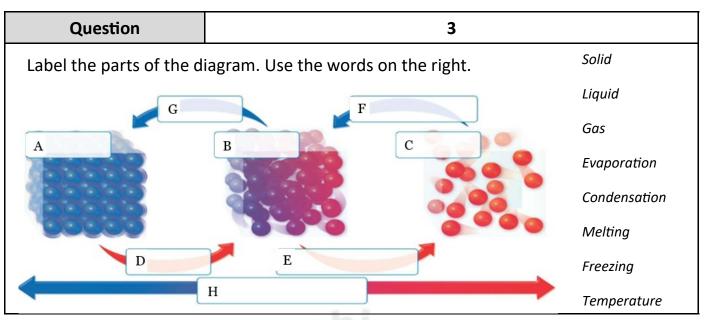
Question				18
Identify by each		atter represented	A	В
А	A—gas B– liquid C– solid		С	no state of matter is represented
В	A—gas B– solid C– liquid		D	A—solid B– liquid C– gas

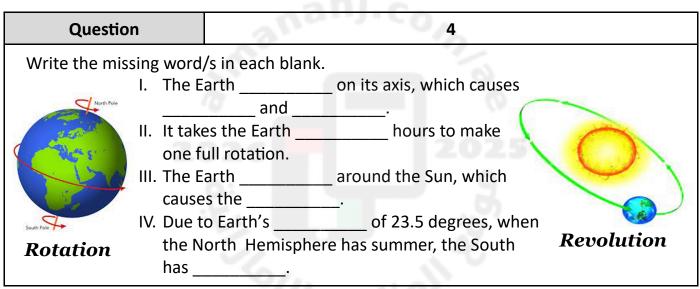
Gr	ade 5	Al Aasimah School—S	Al Aasimah School—Science Mock Exam 2025-2026		
Q	Question 19				
What pulls objects toward Earth's surface?					
А		The wind	С		The moon
В	Th	e sun's heat	D		gravity
0	uestion			20	
•		ighter than others?			
А	All stars are	the same brightness.	С		oon makes some stars shine brighter.
В	Some sta	D	Some stars are bigger and closer to Earth.		
				0)	
PART 2—WRITING					
Question 1					
Identify the following examples as physical or chemical change. I. Frying an egg					
Question 2					
Write the correct property from the list below. Conductivity—Reflectivity—Solubility—Magnetism—Corrosion					
	Exar	nple of Property		Type of the	e Physical Property
The a	bility of matte	er to dissolve in a liquid			
The way light reflects					

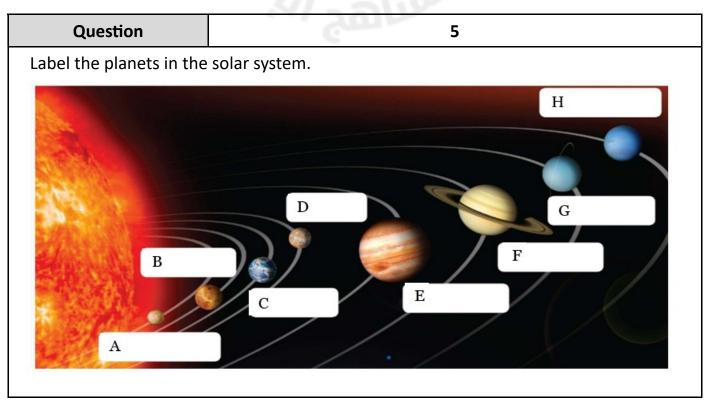
Describes how energy can move through material

The material is attracted to a magnet









ANSWER KEY

MCQs	Writing Part
1. C	1. Changes in Matter
2. A	I. Chemical II. Chemical
3. C	III. Physical IV. Chemical
	V. Chemical
4. A	VI. Physical VII. Chemical
5. C	VII. Physical
6. A	2. Properties
7. C	A. solubility B. reflectivity
7. C	C. conductivity
Ο Λ	D. magnetism
8. A	
9. C	3. Changing States of Matter A. solid
2026	B. liquid
10. A	C. gas
11. D	D. melting
11. D	E. evaporation F. condensation
12. D	G. freezing
\\(\sigma_{11}\),	H. temperature
13. D	
14. D	4. Motion of the Earth I) rotates, day, night
14.2	II) 24
15. D	III) revolves, seasons
46 D	IV) tilt, winter
16. D	5. Planets in the Solar System
17. D	A. Mercury
1/. D	B. Venus
18. D	C. Earth
	D. Mars
19. D	E. Jupiter F. Saturn
00 D	G. Uranus
20.D	H. Neptune